organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Daniel E. Lynch^a* and Ian McClenaghan^b

^aSchool of Science and the Environment, Coventry University, Coventry CV1 5FB, England, and ^bKey Organics Ltd, Highfield Industrial Estate, Camelford, Cornwall PL32 9QZ, England

Correspondence e-mail: apx106@coventry.ac.uk

Key indicators

Single-crystal X-ray study T = 150 KMean σ (C–C) = 0.007 Å H-atom completeness 96% Disorder in main residue R factor = 0.096 wR factor = 0.299 Data-to-parameter ratio = 17.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Diethyl (4-*tert*-butyl-1,3-thiazol-2-ylaminomethylene)malonate

The molecule of the title compound, $C_{15}H_{23}N_2O_4S$, is essentially planar, except for the *tert*-butyl group, with disorder of the aminomethylene atoms. These are equally disordered over two sites each, giving two alternative intramolecular hydrogen bonds from the partial amine H atoms to the adjacent carbonyl O atoms. An additional, intermolecular, $C-H\cdots O$ close contact is observed from the only thiazole ring H atom to one of the carbonyl O atoms. Received 8 January 2003 Accepted 15 January 2003 Online 31 January 2003

Comment

2-Aminothiazole derivatives have a variety of pharmaceutical properties and recent experience has shown us that they are generally easy compounds to crystallize, or cocrystallize. There are currently 157 crystal structures (Cambridge Structural Database, September 2002 release; Allen, 2002) containing a 2-aminothiazole moiety. Only two of these structures have a 4-tert-butyl group, namely 4-tert-butyl-2-(Nmethyl-N-phenylamino)thiazole-5-carbaldehyde (Gillon et al., 1983) and anti-5-acetyl-2-dimethylamino-4-tert-butylthiazole (Caldwell et al., 1987). The crystallographic quality of these two structures is very good, considering the potential for disorder in tert-butyl groups. Ironically, in the title compound, (I), the groups where disorder might be more likely, such as the tert-butyl and/or the ethyl esters, are relatively rigid, compared to the rest of the molecule. Large and very anisotropic displacement ellipsoids for three of the thiazole ring atoms (S1, C2 and N3) and the malonate atoms means that several atoms could be split into two positions each, but attempts to do so resulted in totally unrealistic bond distances and angles. The crystallographic results for (I) are of low precision, and a poor data set was the direct result of poor crystal quality. However, the resolvable disorder in this molecule and the probable unresolved disorder in the atoms with high displacement parameters is worth reporting.



The structure of (I) comprises an essentially planar molecule, except for the *tert*-butyl group, with resolved disorder of the aminomethylene atoms (Fig. 1). All the molecules in the unit cell lie essentially parallel to the *ab* plane. The aminomethylene atoms (N21 and C22) are both equally disordered

© 2003 International Union of Crystallography Printed in Great Britain – all rights reserved over two sites; thus, two alternative intramolecular hydrogen bonds exists from the partial amine H atoms to the adjacent carbonyl O atoms (Table 1). An additional $C-H\cdots O$ close contact is observed from the only thiazole ring H atom to one of the carbonyl O atoms in an adjacent molecule.

Experimental

The title compound was obtained from Key Organics Ltd and crystals were grown from iso-octane.

Crystal data

$C_{15}H_{23}N_2O_4S$ $M_r = 327.41$ Monoclinic, C2/c a = 19.663 (4) Å b = 13.177 (3) Å c = 13.708 (3) Å $\beta = 103.69$ (3)° V = 3450.8 (13) Å ³ Z = 8	$D_x = 1.257 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 7703 reflections $\theta = 2.9-27.5^{\circ}$ $\mu = 0.21 \text{ mm}^{-1}$ T = 150 (2) K Needle, colourless $0.34 \times 0.10 \times 0.08 \text{ mm}$
Data collection	
Bruker–Nonius KappaCCD area- detector diffractometer φ and ω scans Absorption correction: multi-scan (<i>SORTAV</i> ; Blessing, 1995) $T_{\min} = 0.933, T_{\max} = 0.984$ 12347 measured reflections	3916 independent reflections 1830 reflections with $l > 2\sigma(I)$ $R_{int} = 0.113$ $\theta_{max} = 27.5^{\circ}$ $h = -25 \rightarrow 25$ $k = -17 \rightarrow 17$ $l = -16 \rightarrow 17$
Refinement	
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.096$ $wR(F^2) = 0.300$ S = 1.16 3916 reflections	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.1206P)^{2} + 1.7968P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{\text{max}} < 0.001$ $\Delta\rho_{\text{max}} = 0.44 \text{ e } \text{\AA}_{o}^{-3}$

Table 1

226 parameters

Hydrogen-bonding geometry (Å, °).

H-atom parameters constrained

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
N21A-H21A···O28	0.88	2.26	2.931 (10)	133
$N21B - H21B \cdot \cdot \cdot O24$	0.88	2.33	2.913 (9)	123
$C5-H5\cdots O28^i$	0.95	2.44	3.363 (7)	163

 $\Delta \rho_{\rm min} = -0.94 \text{ e} \text{ Å}^{-3}$

Extinction correction: SHELXL97

Extinction coefficient: 0.0067 (16)

Symmetry code: (i) $\frac{1}{2} - x$, $y - \frac{1}{2}, \frac{1}{2} - z$.

All H atoms were included in the refinement, at calculated positions, as riding models, with X-H set to 0.88 (N-H), 0.95



Figure 1

The molecular structure and atom-numbering scheme for the title compound, showing 50% probability ellipsoids and the resolved twofold disorder.

(aromatic), 0.99 (CH₂) or 0.98 Å (CH₃), and $U_{iso}(H) = 1.25U_{eq}(X)$. The high R_{int} value is the result of weak high-angle data.

Data collection: *DENZO* (Otwinowski & Minor, 1997) and *COLLECT* (Hooft, 1998); cell refinement: *DENZO* and *COLLECT*; data reduction: *DENZO* and *COLLECT*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *PLATON*97 (Spek, 1997); software used to prepare material for publication: *SHELXL*97.

We thank the EPSRC National Crystallography Service (Southampton) and acknowledge the use of the EPSRC's Chemical Database Service at Daresbury (Fletcher *et al.*, 1996).

References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Blessing, R. H. (1995). Acta Cryst. A51, 33-37.
- Caldwell, J. M., Meakins, G. D., Jones, R. H., Kidd, T. R. & Prout, K. (1987). J. Chem. Soc. Perkin Trans. 1, pp. 2305–2310.
- Fletcher, D. A., McMeeking, R. F. & Parkin. D. (1996). J. Chem. Inf. Comput. Sci. 36, 746–749.
- Gillon, D. W., Forrest, I. J., Meakins, G. D., Tirel, M. D. & Wallis, J. D. (1983). J. Chem. Soc. Perkin Trans. 1, pp. 341–347.
- Hooft, R. (1998). COLLECT. Nonius BV, Delft, The Netherlands.

Otwinowski, Z. & Minor, W. (1997). Methods in Enzymology, Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr and R. M. Sweet, pp. 307–326. New York: Academic Press.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Spek, A. L. (1997). PLATON97. University of Utrecht, The Netherlands.